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1. Gases consist of tiny particles (atoms or molecules)
2. These particles are so small, compared with the distance between them, that the volume (size) of the individual particles can be assumed to be negligible (zero)
3. The particles

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A gas is a state of matter with no defined shape or volume. Gases have

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their own unique behavior

depending on a variety of variables, such as temperature, pressure, and volume. While each gas is different, all gases act in a similar matter. This study guide highlights the concepts and laws dealing with the chemistry of gases.

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Characteristics of Gases •All

substances have three phases: solid,
liquid and gas •Substances that are
liquids or solids under ordinary
conditions may also exist as gases

•These are often referred to as
vapors •Many of the properties of
gases differ

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substances have three phases: solid,

liquid and gas •Substances that are

liquids or solids under Page 10/26.

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Substances have three phases: solid, liquid and gas. •Substances that are liquids or solids under ordinary conditions may also exist as gases. •These are often referred to as vapors. •Many of the properties of gases differ from those of solids ...

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